

COURSE OUTLINE		
TERM: Spring 2023	COURSE NO: CHEM 111	
INSTRUCTOR:	COURSE TITLE: CHEMICAL DYNAMICS AND ENERGETICS	
OFFICE: LOCAL:	SECTION NO(S):	CREDITS: 4.0
E-MAIL: @capilanou.ca		
OFFICE HOURS:		
COURSE WEBSITE:		

Capilano University acknowledges with respect the Lilwat7úl (Lil'wat), xʷmə̓ ǀ ʰkʷəyəm (Musqueam), shísháhlh (Sechelt), Sk̓wxwú7mesh (Squamish), and Səlílwətaʔ/Selilwitulh (Tsleil-Waututh) people on whose territories our campuses are located.

COURSE FORMAT

Three hours of class time, three lab hours plus an additional hour of supplemental activity delivered through on-line or other activities for a 15-week semester, which includes two weeks for final exams.

COURSE PREREQUISITES

CHEM 110 (C-); and Precalculus 12 (B) or BMTH 054 (C+) or MATH 105 (C-) or Calculus MPT.

CALENDAR DESCRIPTION

This is a course for students who plan to pursue a science degree. Students will review and expand their understanding of stoichiometry and the gaseous state; kinetics and reaction mechanisms; chemical equilibrium; acids, bases, salts and buffers; thermodynamics; solubility; redox and electrochemistry.

COURSE NOTES

CHEM 111 is an approved Science and Technology course for Cap Core requirements.

CHEM 111 is an approved Quantitative/Analytical course for baccalaureate degrees.

CHEM 111 is an approved Science course.

CHEM 111 is an approved Laboratory Science course.

REQUIRED TEXTS AND/OR RESOURCES

Textbook: Chang, R. and Overby, J. *Chemistry*. 14th ed. Toronto: McGrawHill, 2019.
Cusanelli, T. and Smith, J. *CHEM 110/111 Organic Chemistry*. Custom publication. Toronto: McGraw-Hill, 2022.
Capilano University Chemistry 111 Laboratory Manual
Capilano University Student Laboratory Guide for First Year Chemistry.

Supplement: Organic Chemistry Model Kit is recommended
Graphing calculator (TI-83 or equivalent)

COURSE STUDENT LEARNING OUTCOMES

On successful completion of this course, students will be able to do the following:

- Balance chemical equations and solve stoichiometric problems in the solid, solution and gas phase.
- Solve kinetics problems and relate experimental kinetic evidence to reaction mechanisms.
- Solve a variety of equilibria problems, including acids and bases, buffers, salts and solubility.
- Apply the principles of thermodynamics and thermochemistry to chemical reactions.
- Explain the basic concepts of electrochemistry and how it relates to batteries, corrosion, etc.

Students who complete this Science and Technology course will be able to do the following:

- Apply numerical and computational strategies to solve problems.
- Demonstrate how a problem, concept, or process can be modelled numerically, graphically, or algorithmically.
- Explain how scientific inquiry is based on investigation of evidence and evolves based on new findings.
- Participate in scientific inquiry and communicate the elements of the process, including making careful and systematic observations, developing and testing a hypothesis, analysing evidence, and interpreting results.

COURSE CONTENT

Topic	Weeks (approx)
Introductory Material, Stoichiometry and Gases Most of the material in these chapters should be review and includes the chemical language, basic atomic structure, dimensional analysis, isotopes, the mole, properties of solutions, types of chemical reactions & stoichiometry, limiting reactants, the gas laws and their applications.	1-2
Chemical Kinetics and Organic Chemistry <ul style="list-style-type: none"> • An introduction to reaction rate, the rate law and reaction order, activation energy and temperature, mechanisms and catalysis 	3-5
Chemical Equilibrium <ul style="list-style-type: none"> • The concept of chemical equilibrium, gas equilibrium, the relationship between equilibrium and kinetics, LeChatelier's principle and solving equilibrium problems 	6
Acids, Bases and Solubility Equilibria <ul style="list-style-type: none"> • General properties of acids and bases such as the pH scale, the ion-product constant of water, Bronsted-Lowry and Lewis definitions of acids and bases • Weak acids and bases, di- and polyprotic acids, salt hydrolysis, the common ion effect, buffer solutions, acid-base titrations and indicators • The solubility product, fractional precipitation, qualitative analysis 	7-10
Chemical Thermodynamics	11-12

Topic	Weeks (approx)
<ul style="list-style-type: none"> Energy changes, enthalpy, calorimetry, Hess's Law, bond energies, first and second law, entropy, Gibbs free energy, relation of thermodynamics to equilibrium 	
Electrochemistry <ul style="list-style-type: none"> Definitions, oxidation number, balancing redox equations, galvanic cells, standard potentials, the thermodynamics of electrochemical reactions, Nernst equation, concentration cells, batteries, fuel cells and corrosion 	13
Final Exam Period	14-15

EVALUATION PROFILE

Final grades for the course will be computed based on the following schedule:

Term Work	40%
Laboratory Work	20%
Performance Evaluation	5%
Final Examination	35%
TOTAL	100%

Term work may consist of tests, quizzes and/or assignments. No single component of term work will be worth more than 25%. Laboratory work will consist of quizzes, laboratory reports and other assessments. The weight and format of individual tests, quizzes, and assignments, etc. are assigned by the instructor and will be announced in class in advance.

A pass grade of 50% or above is required on each of the laboratory and term work portions of the course for the student to pass the course.

Specific dates and details regarding the Evaluation Component will be provided by the instructor.

Performance Evaluation

In the absence of exceptional circumstances, which are evaluated at the instructor's discretion, the performance evaluation component of the final grade will be pro-rated to the rest of the grade. For example, a 10% performance evaluation component would be determined by dividing the remaining mark out of 90 by 9. The most common circumstance justifying an increased performance evaluation mark is a student's improved performance in the final examination relative to the term work, which the instructor feels justifies an elevated letter grade.

GRADING PROFILE

A+ 90 - 100%	B+ 77 - 79%	C+ 67 - 69%	D 50 - 59%
A 85 - 89%	B 73 - 76%	C 63 - 66%	F 0 - 49%
A- 80 - 84%	B- 70 - 72%	C- 60 - 62%	

Students should refer to the University Calendar for the effect of the above grades on grade point average.

Incomplete Grades

Grades of Incomplete "I" are assigned only in exceptional circumstances when a student requests extra time to complete their coursework. Such agreements are made only at the request of the student, who is responsible to determine from the instructor the outstanding requirements of the course.

Late Assignments

Assignments are due at the beginning of the class on the due date listed. If you anticipate handing in an assignment late, please consult with your instructor beforehand.

Missed Exams/Quizzes/Labs etc.

Students must inform their instructor on the day of the exam or beforehand, if they are unable to attend. Make-up exams, tests and/or labs or extensions on assignment due dates are given at the discretion of the instructor. They are generally given only in medical emergencies or severe personal crises, and to students who have been fully participating in the course until that time. Some missed labs or other activities may not be able to be accommodated. Please consult with your instructor.

***Accommodations can be made to honour community needs and traditional practices.

Attendance

Students are expected to attend all classes and associated activities. If classes are missed, it is the student's responsibility to become aware of all information given out in the classes or tutorials, including times of examinations and assignment deadlines.

English Usage

Students are expected to use correct standard English in their written and oral assignments, exams, presentations, and discussions. Failure to do so may result in reduced grades in any part of the Evaluation Profile. Please refer to the guidelines provided in the Capilano Guide to Writing Assignments (available from the University Bookstore).

Electronic Devices

Students may use electronic devices during class; however, an instructor may ask for devices to be put away if they become a distraction to other students or interfere with classroom learning.

Online Communication

Outside of the classroom, instructors will (if necessary) communicate with students using either their official Capilano University email or eLearn; please check both regularly. Official communication between Capilano University and students is delivered to students' Capilano University email addresses only.

Students are reminded to engage in respectful behavior when participating in discussions.

UNIVERSITY OPERATIONAL DETAILS:

Tools for Success

Many services are available to support student success for Capilano University students. A central navigation point for all services can be found at: <https://www.capilanou.ca/student-life/>

Capilano University Security: download the [CapU Mobile Safety App](#)**Policy Statement (S2009-06)**

Capilano University has policies on Academic Appeals (including appeal of final grade), Student Conduct, Cheating and Plagiarism, Academic Probation and other educational issues. These and other policies are available on the University website.

Academic Integrity (S2017-05)

Any instance of academic dishonesty or breach of the standards of academic integrity is serious and students will be held accountable for their actions, whether acting alone or in a group. See policy and procedures S2017-05 for more information: <https://www.capilanou.ca/about-capu/governance/policies/>

Violations of Academic Integrity, including dishonesty in assignments, examinations, or other academic performances, are prohibited and will be handled in accordance with the Student Academic Integrity Procedures.

Academic dishonesty is any act that breaches one or more of the principles of academic integrity. Acts of academic dishonesty may include but are not limited to the following types:

Cheating: Using or providing unauthorized aids, assistance or materials while preparing or completing assessments, or when completing practical work (in clinical, practicum, or lab settings), including but not limited to the following:

- Copying or attempting to copy the work of another during an assessment;
- Communicating work to another student during an examination;
- Using unauthorized aids, notes, or electronic devices or means during an examination;
- Unauthorized possession of an assessment or answer key; and/or,
- Submitting of a substantially similar assessment by two or more students, except in the case where such submission is specifically authorized by the instructor.

Fraud: Creation or use of falsified documents.

Misuse or misrepresentation of sources: Presenting source material in such a way as to distort its original purpose or implication(s); misattributing words, ideas, etc. to someone other than the original source; misrepresenting or manipulating research findings or data; and/or suppressing aspects of findings or data in order to present conclusions in a light other than the research, taken as a whole, would support.

Plagiarism: Presenting or submitting, as one's own work, the research, words, ideas, artistic imagery, arguments, calculations, illustrations, or diagrams of another person or persons without explicit or accurate citation or credit.

Self-Plagiarism: Submitting one's own work for credit in more than one course without the permission of the instructors, or re-submitting work, in whole or in part, for which credit has already been granted without permission of the instructors.

Prohibited Conduct: The following are examples of other conduct specifically prohibited:

- Taking unauthorized possession of the work of another student (for example, intercepting and removing such work from a photocopier or printer, or collecting the graded work of another student from a stack of papers);
- Falsifying one's own and/or other students' attendance in a course;
- Impersonating or allowing the impersonation of an individual;
- Modifying a graded assessment then submitting it for re-grading; or,
- Assisting or attempting to assist another person to commit any breach of academic integrity.

Sexual Violence and Misconduct

All Members of the University Community have the right to work, teach and study in an environment that is free from all forms of sexual violence and misconduct. Policy B401 defines sexual assault as follows:

Sexual assault is any form of sexual contact that occurs without ongoing and freely given consent, including the threat of sexual contact without consent. Sexual assault can be committed by a stranger, someone known to the survivor or an intimate partner.

Safety and security at the University are a priority and any form of sexual violence and misconduct will not be tolerated or condoned. The University expects all Students and Members of the University Community to abide by all laws and University policies, including B.401 Sexual Violence and Misconduct Policy and B.401.1 Sexual Violence and Misconduct Procedure (found on Policy page <https://www.capilanou.ca/about-capu/governance/policies/>)

Emergency Procedures

Students are expected to familiarise themselves with the emergency policies where appropriate and the emergency procedures posted on the wall of the classroom.

DEPARTMENT OR PROGRAM OPERATIONAL DETAILS:

Professionalism

Students are expected to demonstrate a professional attitude and behaviour: reliability, respect for and cooperation with colleagues, willingness to work calmly and courteously, respect for equipment and systems, and constructive response to criticism.

Final Exam Period

Students should note the dates of final exam period and expect to write exams at any time during this period. Individual exam times will not normally be rescheduled because of holidays, work, or other commitments. While efforts are made to spread exams throughout the exam period, an individual's particular course combination may result in exams being scheduled close together or spread widely through the entire exam period.